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Chemical Equations
and Reactions MIXED
REVIEW SHORT

ANSWER Answer the following questions in the space provided. 1. b A balanced chemical equation represents all the following except (a)

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experimentally established facts. (b) the mechanism by which reactants combine to form products.

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Cracking the AP
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the AP Chemistry
Exam. Chapter 8. Big
Idea #6: Equilibrium,
Acids and Bases,
Titrations, and
Solubility. CHAPTER 8
ANSWERS AND
EXPLANATIONS.

Multiple-Choice. 1. A
The equivalence point
is defined as the point
at which the moles of
acid is equal to the
moles of base. The
moles of NaOH is equal
to $(1.0 \text{ M})(0.0200$
 $\text{L})=0.0200 \text{ mol.}$

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CHAPTER 8
ANSWERS AND
EXPLANATIONS -
Equilibrium, Acids ...

wstobbs. Modern
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chemical equation.
skeleton equation.
balanced equation.
precipitate. a concise
representation of a
chemical reaction
using chemical.... a
chemical equation that
does not indicate the
relative amount.... an

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equation in which each side has the same number of atoms of....

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1. What information does a molecule's structure give? 2. How many resonance structures can be drawn for CO_3^{2-} ? Show the electron dot structures for each. 3. How is an

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intermolecular force
different from a bond?

4. What types of
molecules have the
following
intermolecular force a.)
Dispersion Force b.)
Dipole Interaction c.)
Hydrogen Bonding 5.
Would you expect
carbon monoxide Co
and carbon ...

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**Questions: Chapter
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Answers

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8 Chapter 1 Review
Solutions 9. It does not.
A significant figure is
one that has been
measured. It doesn't
have to be

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mathematically important. For example, the zero in 1.0 is significant, but it is not mathematically important, as 1.0 and 1 have the same mathematical value. 10. There is always uncertainty in the last significant figure.

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