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The element mercury, Hg, is a silvery liquid at room temperature. The normal freezing point of mercury is -38.9°C , and its molar enthalpy of fusion is $\Delta H_{\text{fusion}} = 2.29 \text{ kJ/mol}$. What is the entropy change of the system when 50.0 g of Hg(l) freezes at the normal freezing

Chapter 19 Chemical Thermodynamics

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Created by. Mackenzie_Koenig. Terms in this set (40) acidic solution. Contains more hydrogen ions than hydroxide ions. basic solution. Any solution in which the hydroxide-ion concentration is greater

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mass of the reactants and the mass of
the products in a chemical reaction, and
explain how these masses are related.
The coefficients in the balanced
equation indicate the molar relationship
between each ...

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